

Topological Effects and Binding Modes Operating with Multivalent Iminosugar-Based Glycoclusters and Mannosidases

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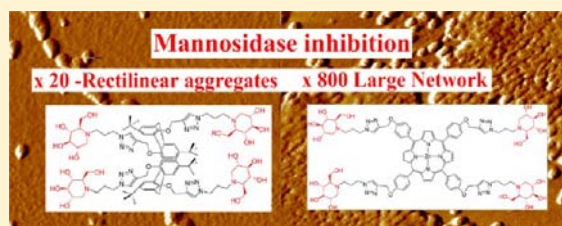
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Supporting Information

ABSTRACT: Multivalent iminosugars have been recently explored for glycosidase inhibition. Affinity enhancements due to multivalency have been reported for specific targets, which are particularly appealing when a gain in enzyme selectivity is achieved but raise the question of the binding mode operating with this new class of inhibitors. Here we describe the development of a set of tetra- and octavalent iminosugar probes with specific topologies and an assessment of their binding affinities toward a panel of glycosidases including the Jack Bean α -mannosidase (JB α Man) and the biologically relevant class II α -mannosidases from *Drosophila melanogaster* belonging to glycohydrolase family 38, namely Golgi α -mannosidase ManIIb (GM) and lysosomal α -mannosidase LManII (LM). Very different inhibitory profiles were observed for compounds with identical valencies, indicating that the spatial distribution of the iminosugars is critical to fine-tune the enzymatic inhibitory activity. Compared to the monovalent reference, the best multivalent compound showed a dramatic 800-fold improvement in the inhibitory potency for JB α Man, which is outstanding for just a tetravalent ligand. The compound was also shown to increase both the inhibitory activity and the selectivity for GM over LM. This suggests that multivalency could be an alternative strategy in developing therapeutic GM inhibitors not affecting the lysosomal mannosidases. Dynamic light scattering experiments and atomic force microscopy performed with coincubated solutions of the compounds with JB α Man shed light on the multivalent binding mode. The multivalent compounds were shown to promote the formation of JB α Man aggregates with different sizes and shapes. The dimeric nature of the JB α Man allows such intermolecular cross-linking mechanisms to occur.



INTRODUCTION

Synthetic glycoclusters presenting multiple copies of a sugar epitope have been extensively investigated to inhibit carbohydrate binding proteins called lectins.¹ Logarithmic affinity enhancements compared to monovalent references are often observed with fine-tuned glycoclusters. These affinity improvements termed “multivalent or glycoside cluster effects”² can be rationalized by a limited number of binding mechanisms occurring independently or simultaneously.³ The highest multivalent effects have been reported when the tethered epitopes of a glycocluster embrace at least two binding sites of a

multimeric lectin (chelate binding mode).⁴ When the linkers are too short for such chelation, organized and stabilized glycocluster–lectin networks can be formed. Both interactions require a multimeric presentation of the receptors, a situation generally encountered with lectins.

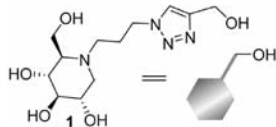
In stark contrast, multivalency was largely disregarded as a means to inhibit carbohydrate processing enzymes, until recently. Only a few examples of multivalent glycosidase

Received: July 8, 2013

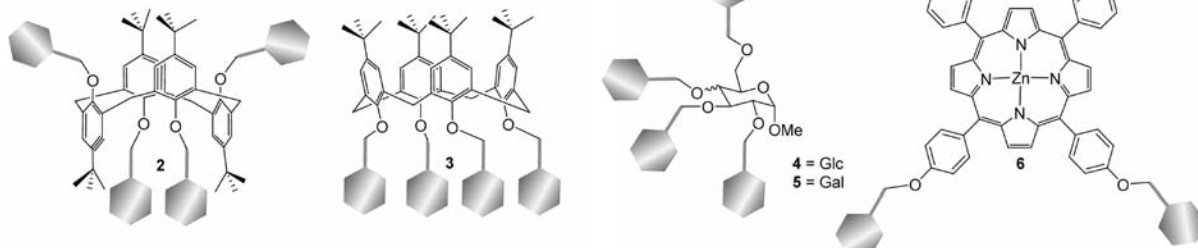
Published: November 13, 2013

Chart 1. Structure of the Multivalent Glycoclusters Incorporating 4–8 DNJ Moieties

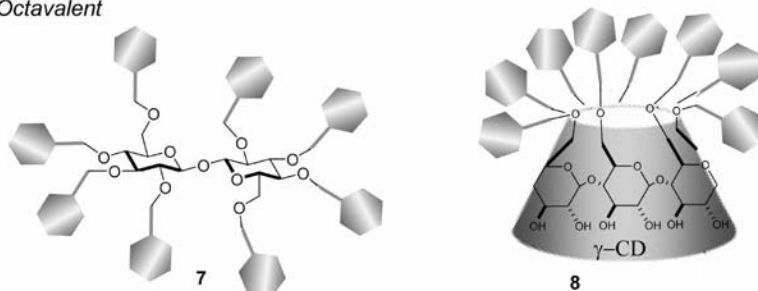
Monovalent reference



Tetravalent



Octavalent



inhibitors were reported in the literature.⁵ This can be explained by the synthetic hurdles in obtaining acceptable quantities of inhibitors for multivalent grafting and by the generally monovalent nature of glycosidases, which are not prone to chelate or aggregative processes. Nevertheless, significant affinity enhancements may also be expected with multivalent conjugates interacting at a single protein-binding site. The locally high concentration of the tethered ligands in close proximity to the binding site may favor a rebinding mechanism and prevent complex dissociation, as previously observed with lectins.⁶ These considerations stimulated us to study multivalency in the context of glycosidase inhibition. In previous work, we designed mono-, di-, and trivalent deoxyribose (DNJ) analogues (a broad glycosidase inhibitor) grafted onto ethylene glycol scaffolds by click chemistry and assessed the inhibitory activities of our compounds toward a panel of glycosidases. We observed different inhibitory profiles depending on the enzymes and showed the first significant multivalent effect on a glycosidase (Jack Bean- α -D-mannosidase = JB α Man).⁷ These results were further supported by others with click DNJ-based dodecavalent fullerenes showing high multivalent effects (up to 3 orders of magnitude) on the same enzyme.⁸ Multivalent effects were also reported on biologically relevant enzymes such as glucosylceramide synthase and the bacterial heptosyltransferase WaaC, with implications for Gaucher diseases⁹ and bacterial infections,¹⁰ respectively. Together, these results highlight the great potential for multivalent inhibition of glycosidase enzymes and pave the way to potential biomedical applications.

Surprisingly, positive responses to multivalency are only observed with certain enzymes, which is very beneficial for selectivity improvements but remains intriguing. Insights into the multivalent binding mode operating should give clues about

such selectivity and may enable prediction of the efficiency of multivalent inhibitors for new targets. In the present work, we wished to assess the multivalent binding mode(s) operating with JB α Man, for which the first and the highest multivalent effect has been described. JB α Man is a class II- α -mannosidase, a group of glycoside hydrolases of considerable interest. Golgi α -mannosidase (GM) and lysosomal α -mannosidases (LM) are two other members of this family (GH38) with closely related protein sequences. Inhibition of the former enzyme by the indolizidine alkaloid swainsonine has been shown to reduce tumor growth and cell metastasis,¹¹ while a deficiency in the latter is observed in the inherited disorder mannosidosis. There is a significant interest in developing selective GM inhibitors over LM to prevent unwanted mannosidosis syndromes during cancer chemotherapies.

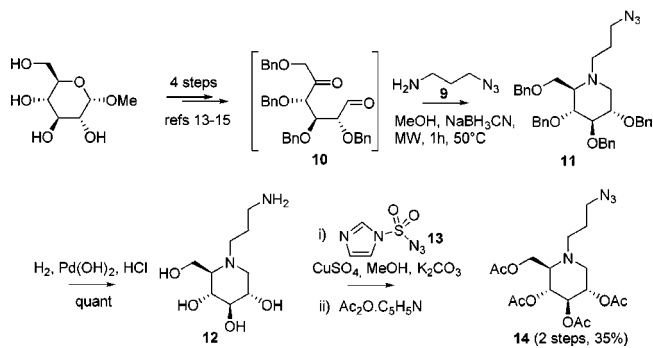
On the basis of structural analogies, JB α Man is considered a good model for these two glycosidases, and studying multivalent effects on JB α Man to develop a new enzymatic mode of inhibition is therefore of particular interest. For this purpose, we designed a series of mono-, tetra-, and octavalent DNJ probes 1–8 based on porphyrin, calix[4]arene, glucose, galactose, trehalose and γ -cyclodextrin scaffolds (Chart 1).

The different scaffolds selected give specific spatial orientations of the tethered DNJ, and may thus impact the inhibition. The inhibitory activities of the compounds were tested on a range of commercially available glycosidases including JB α Man. The two compounds showing the highest inhibition of JB α Man were also tested against GM and LM to evaluate the potential of this multivalent approach in anticancer therapy.

Dynamic light scattering experiments (DLS) and atomic force microscopy (AFM) images shed light on the binding mode operating with the multivalent iminosugars and JB α Man.

Synthesis. Benzylated azido-functionalized DNJ **11**⁷ was obtained from commercially available methyl- α -D-glucopyranoside as we previously described (Scheme 1).

Scheme 1. Synthesis of the Acetylated Azido-Functionalized DNJ Intermediate 14



This procedure requires a critical double-reductive amination step involving 3-azidopropylamine **9**¹³ and the 1,5-dicarbonylated compound **10**.¹⁴ Deprotection of the benzyl groups was problematic on octavalent and porphyrin-based compounds. Thus, we synthesized a second *N*-azidopropyl-DNJ synthon **14** with acetate protecting groups. The double-reductive amination

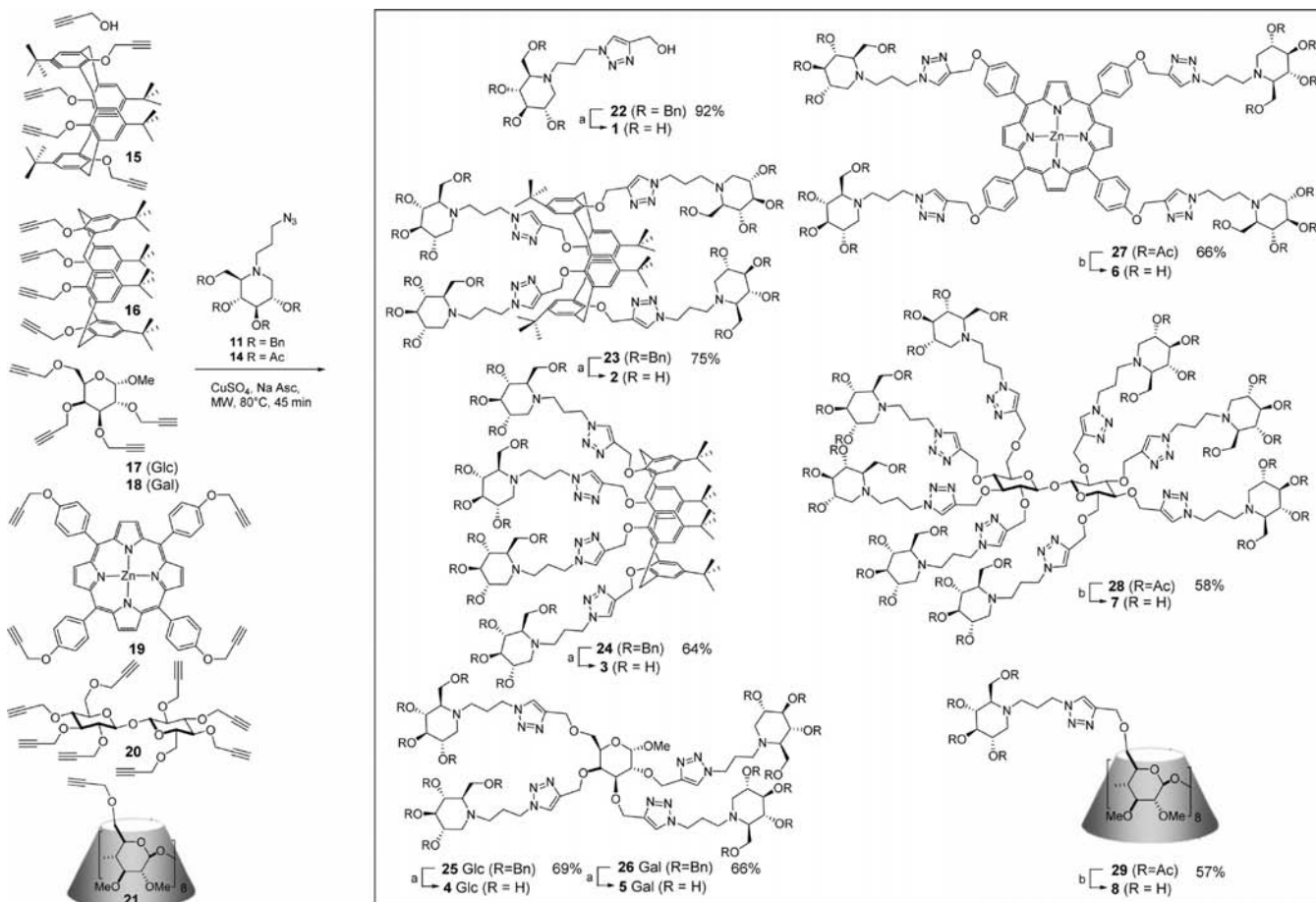
developed for the cyclization of **11** was not applicable to compound **14** due to the rather labile esters under these conditions. Thus, we chose to obtain compound **14** directly from **11**. Initial attempts to remove the benzyl groups selectively with BCl_3 failed and gave a complex mixture of products.¹⁵ Azides and benzyl groups of **11** were then reduced with Pearlman's catalyst in acidic media.¹⁶ The azide group was recovered with a copper-catalyzed diazo-transfer reaction performed with imidazole-1-sulfonyl azide **13**,¹⁷ a safer reagent than the generally used triflyl azide.¹⁸ The crude product was acetylated in a mixture of pyridine and acetic anhydride to give **14**.

Propargylated scaffolds **15**,¹⁹ **16**,²⁰ **17**,²¹ **18**, **20**, and **21** (Scheme 2) were obtained in yields ranging from 46 to 90% with propargyl bromide, sodium hydride, and the corresponding alcohols in DMF (see the Supporting Information). Compound **19** was obtained as previously described.²²

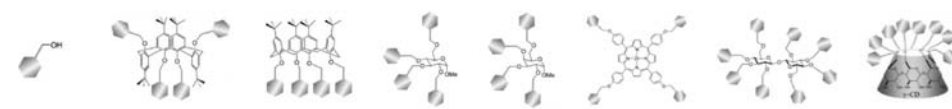
The protected azidopropyl-DNJ **11** or **14** was tethered to the propargylated scaffolds **15**–**21** with microwave-assisted copper-catalyzed azide-alkyne cyclization (CuAAC). The compounds were irradiated in the presence of copper sulfate and sodium ascorbate in a mixture of dioxane and water.

The exclusive formation of the 1,4-cycloadducts was observed in all cases as shown by the large ($\delta\text{C-4}-\delta\text{C-5}$) values (>20 ppm) recorded by ¹³C NMR spectroscopy for the different structures.²³ The 1,3-dipolar cycloaddition was initially

Scheme 2. Structure of the Eight Glycoclusters Obtained by CuAAC^a



^aYields are indicated for the cyclization-deprotection steps. Conditions: (a) $\text{Pd}(\text{OH})_2$, MeOH –1 M aq HCl; rt, 48 h; (b) Amberlite resin IRN 78, MeOH – H_2O , rt, 14 h.

Table 1. Glycosidase Inhibitory Activities (K_i , μM) for Compounds 1–8^a


	1	2	3	4	5	6	7	8
β -galactosidase (bovine liver)	n.i.	58 \pm 6	38 \pm 4	n.i.	64 \pm 6	52 \pm 5	71 \pm 9	191 \pm 19
β -galactosidase (E. coli)	85 \pm 9	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
α -galactosidase (green coffee)	55 \pm 6	36 \pm 4	24 \pm 2	228 \pm 23	24 \pm 2	18 \pm 1	1.7 \pm 0.2	40 \pm 4
β -glucosidase (almonds pH 7.3)	233 \pm 24	729 \pm 87	n.i.	311 \pm 36	283 \pm 28	n.i.	281 \pm 27	177 \pm 18
α -glucosidase (baker's yeast)	n.i.	45 \pm 4	585 \pm 59	n.i.	n.i.	n.i.	243	n.i.
α -mannosidase (jack bean)	400 \pm 48	20 \pm 2	1.5 \pm 0.2	76 \pm 8	41 \pm 4	0.5 \pm 0.05	21 \pm 2	23 \pm 2
β -mannosidase (helix pomatia)	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.	n.i.
amylogucosidase (Aspergillus niger)	45 \pm 4	11	43 \pm 5	11 \pm 1	2.3 \pm 0.2	24 \pm 2	4.6 \pm 0.5	8 \pm 0.8

^an.i. = no inhibition observed at 2 mM concentration of the inhibitor. Most of the compounds displayed significant multivalent effects relative to 1 on α -mannosidase from Jack Bean (gray cells).

performed with 11 and propargyl alcohol to form the monovalent reference 1 after hydrogenolysis of the benzyl groups of 22 with Pearlman's catalyst in acidic media. Deprotection of the benzyl groups by such protocols was shown to be problematic with octavalent and porphyrin-based scaffolds and incomplete on the octavalent iminosugars. This problem was overcome with the corresponding acetylated analogues 27–29 that were easily deprotected using basic anion-exchange resin Amberlite IRN-78. Compounds 1–9 were obtained with fair to good yields (57–92%) for the cycloaddition–deprotection sequence depending on the difficulty of purification using silica gel column chromatography.

Inhibition Studies on Glycosidases. Variation of the spatial ligand presentation on glycoclusters was previously shown to affect the binding profiles for specific lectins.²⁴ Multivalent glycoconjugates with identical valencies were shown to discriminate between lectins with closely related sequences.²⁵ The set of multivalent DNJ 2–6 and 7, 8 based on different chemical architectures but with identical valencies are interesting tools to investigate if such topological effects are also of importance for inhibition. In particular, the alternate 2 and cone calix[4]arene 3 are isomers with distinct spatial presentations of the DNJ epitopes.^{26,27} The two sets of carbohydrate-based ligands 4, 5 and 7, 8 also possess rather similar scaffolds but should provide distinct presentations of the DNJ ligands.

The inhibition constants (K_i) for the monovalent reference 1, the tetravalent compounds 2–6 and the octavalents 7, 8 are presented in Table 1.

In agreement with previous studies,^{7–9} different inhibitory profiles were observed depending on the glycosidase targets. This further supports the hypothesis that multivalency cannot









be considered a general strategy for glycosidase inhibition as it is for lectins since not all enzymes were inhibited by these multivalent inhibitors.

Significant affinity enhancements were observed for JB α Man and multivalent ligands 2–8. In order to quantify better the level of the multivalent inhibitions observed, the relative inhibitory potencies (RIP) and affinity enhancement per DNJ motif (AED) are presented in Table 2.

RIP values are calculated by dividing the K_i of the monovalent reference 1 by the K_i of the multi-DNJ and give the affinity enhancement per molecule. The RIP values were divided by the ligand valency to assess strictly the affinity gain per DNJ motif (positive multivalent effect for AED > 1). All of the compounds showed positive multivalent effects on JB α Man. The cone calix[4]arene 3 showed a strong multivalent effect on JB α Man (RIP = 267) but was surpassed by 6 (RIP = 800) for which a dramatic AED gain of 200 was observed. This tetravalent porphyrin-based compound displayed a higher enhancement per DNJ motif than a previously described dodecavalent DNJ-based fullerene (AED = 179),⁸ which is remarkable for such a low-valency ligand.

JB α Man is a zinc metalloenzyme which can be significantly inactivated by micromolar concentrations of copper.²⁹ We therefore ascertained that the strong affinity observed with the multivalent DNJ was not due to the presence of residual copper from the CuAAC step. The residual concentration of free copper was estimated with a colorimetric test kit (excluding colored compound 6).³⁰ In each case, the residual amount of copper ranged from 50 to 150 ppm. As an example, for the second most potent compound 3 (K_i = 1.5 μM , Cu < 150 ppm) this corresponds to a subnanomolar concentration of copper (Cu < 0.2 nM) during the enzymatic assay, well below the micromolar concentration required for JB α Man inhibition. The

Table 2. Affinity Enhancements Observed with Glycoclusters 2–8 on the α -Mannosidase from Jack Bean

	Compound	Valency	α -mannosidase	
			RIP ^[a]	AED ^[b]
	1	1	1	1
	2	4	20	5
	3	4	267	67
	4	4	5	1.2
	5	4	10	2.5
	6	4	800	200
	7	8	19	2.4
	8	8	17	2.1

^aRIP = relative inhibitory potency calculated by dividing K_i of reference 1 by the K_i of the corresponding multivalent compound. ^bAED = affinity enhancement per DNJ motifs calculated by dividing RIP values by the compound valency. AED values > 1 are indicative of a positive multivalent effect. The highest multivalent effect observed on JB α Man is presented in the gray cells.

glycosidase inhibitory activities observed for the compounds are therefore not related to or affected by the low residual amount of copper.

The octavalent γ -cyclodextrin-based compound **8** only displayed a limited multivalent effect (AED = 2.1), which was surprising considering that β -CD analogues with closely related sequences and spatial distribution of the epitopes were previously shown to be potent inhibitors of the same enzyme.³¹ The latter differed in valency (7- or 14-valent) and displayed more carbon atoms in the *N*-alkyl chain (*N*-hexyl or *N*-nonyl DNJ). Subtle differences in ligand presentations or nonspecific positive interactions of the β -CD scaffold with the mannosidase may explain the strong differences in inhibitory activities among these compounds.

Importantly, topological effects were shown here to influence strongly the selectivity and inhibition for specific glycosidases. This is well illustrated with isomeric alternate **2** and cone calix[4]arene **3** displaying significantly different AED values on JB α Man of 5 and 67, respectively.

These results clearly highlight the importance of the scaffold selection in fine-tuning the epitope presentation in order to increase multivalent glycosidase inhibition. The fact that compounds with the highest valencies are not necessarily the most potent inhibitors may be similar to the situation encountered with lectins, where plateaus of inhibition are often observed when the valency of inhibitors is further increased.³²

Inhibition Studies on GM and LM. Potent GM inhibitors such as the indolizidine alkaloid swainsonine have a strong potential in cancer therapy by blocking *N*-glycan biosynthesis.

However, their development is greatly hampered by unwanted coinhibition of the closely related mannosidase LM, resulting in symptoms of α -mannosidosis.

In view of the results obtained on JB α Man, we wanted to evaluate here if the multivalent strategy could be a means to increase the affinity and selectivity for GM over LM. For this reason we used two *Drosophila melanogaster* enzymes, ManIIb (labeled here GM), a Golgi-localized α -mannosidase, and LManII (labeled here LM), a homologue of human lysosomal α -mannosidases. The recombinant enzymes were expressed in *Pichia pastoris*, processed and assayed as already described.³³ Briefly, concentrated and dialyzed induced culture medium was used as an enzyme in the assay with *p*-nitrophenyl- α -D-mannopyranoside (pNP-Man) as a substrate. Liberated *p*-nitrophenol was measured spectrophotometrically.

The tetravalent inhibitors **3** and **6** showing the highest inhibition of JB α Man, and the monovalent references **1** were assessed on GM and LM (Table 3). Interestingly, an affinity

Table 3. Mannosidase Inhibitory Activities (K_i , μ M) for Compounds 1, 3, and 6^a

compd	GM (K_i/μ M)	LM (K_i/μ M)
1	362 \pm 88	n.i.
3	n.i.	n.i.
6	24 \pm 10	n.i.

^an.i. = no inhibition observed at 1 mM concentration of the inhibitor.

enhancement due to valency was observed with **6**. Although the effect on GM was moderate compared to the inhibition observed on JB α Man, the results show that multivalent iminosugars may be considered to increase affinity for this pharmaceutical target. Compound **6** was also selective for GM with no inhibition observed on LM at the high concentration of 1 mM. At present, the reason for such selectivity is unclear. A specific residue in LM may sterically prevent the approach of the large scaffolds of **3** and **6**. Alternatively, LM may also respond to multivalency but at a smaller level than GM and the associated increased inhibitory activity may remain undetectable (above 1 mM).

Compound **3** did not show any significant GM and LM inhibition which is surprising in view of the inhibitory activity observed on JB α Man. It seems that the multivalent iminosugars are able to discriminate glycosidases from the same family with high homology.

Binding Mode Operating with Jack Bean α -Mannosidase. The fact that multivalent effects only occur with a limited number of glycosidases is of great interest in enzyme selectivity but raises the question of the binding mode(s) operating. It is of particular importance to answer this question in order to rationalize the future development of inhibitors and potentially predict possible multivalent effects with specific targets.

We wanted to assign the binding mode operating with JB α Man for which the highest multivalent effects are reported. JB α Man is a 230 kDa heterodimer with two subunits of around 49 and 66 kDa. The larger subunit was postulated as the catalytically competent active site.³⁴ Interestingly, a recent study suggested that each unit of 115 kDa (49 + 66) was catalytically active in the dimeric structure.³⁵ On the basis of these considerations, four different multivalent binding modes can be proposed to explain the affinity enhancement observed with the multivalent iminosugars. These specific interactions (Figure 1) are frequently observed with carbohydrate binding

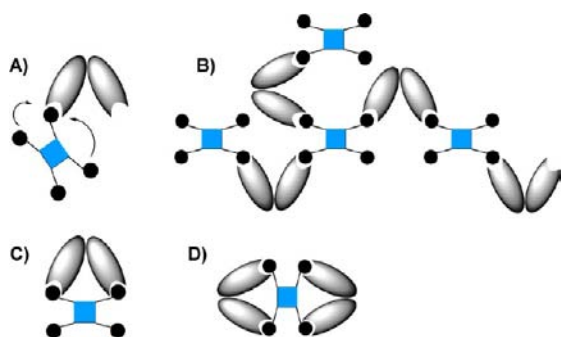


Figure 1. Illustration of the possible binding modes operating with a tetravalent ligand and a divalent protein. (A) recapture mechanism favored by the high local concentration of ligands; (B) formation of ligand-protein networks; (C) chelate binding mode; (D) aggregative chelate.

proteins.^{1a} In the “bind and jump” process (Figure 1A),^{6a,36,37} the high local concentration of the DNJ ligands in close proximity to the binding site favor a recapture mechanism on the adjacent ligands. Although strong affinity enhancements have been reported with lectin and polymeric structures by this mechanism,³⁸ size-controlled glycoclusters of low valency generally do not exceed a 10-fold increase in affinity compared to monovalent ligands. It is therefore unlikely that the strong multivalent effect observed with **6** could be solely due to this recapture mechanism. Aggregative processes (Figure 1B) can be expected with the tetravalent ligands because of the dimeric nature of the enzyme. Different types of ligand–lectin network have been previously described, with moderate to high stabilizing effects.³⁹ Chelate and aggregative chelate (Figures 1C,D) binding modes may be operating if the distance between the tethered iminosugars can embrace both catalytic sites of the divalent glycosidase. The highest multivalent effects reported on lectins with glycoconjugates occurred through chelate interactions.⁴

Atomic force microscopy experiments were performed to shed light on the binding mode operating with the multivalent iminosugars and JB α Man. AFM is a powerful tool for measuring and imaging interactions in biological samples. This technique provides nanoscale views of various important biological processes and structures such as receptor/ligand interactions, cell functions, microbial activity, lipid membrane properties and protein conformation/assemblies. Experiments were performed in triplicate and similar images were recorded for a given compound. AFM imaging of JB α Man alone (see Figure S7, Supporting Information) or coincubated with the monovalent reference **1** (Figure 2) gave isolated spots at the mica surface matching the size of the JB α Man (height of 5.1 ± 0.3 nm). In stark contrast, the interaction between JB α Man and the multivalent compounds **2–8** produced nanoassemblies (Figure 2 and Figures S8 and S9, Supporting Information).

This clearly illustrates that the multivalent iminosugars **2–8** have the ability to aggregate JB α Man. The stabilization effect observed with the multivalent iminosugars may therefore be related to the formation of the stabilized supramolecular aggregates, in analogy to that previously described for lectin-saccharide networks.³⁸

Remarkably, aggregates with different sizes and shapes were observed in AFM images depending on the compounds. Alternate calix[4]arene **2** formed rectilinear structures with JB α Man (Figure 2B). The specific presentation of the DNJ in

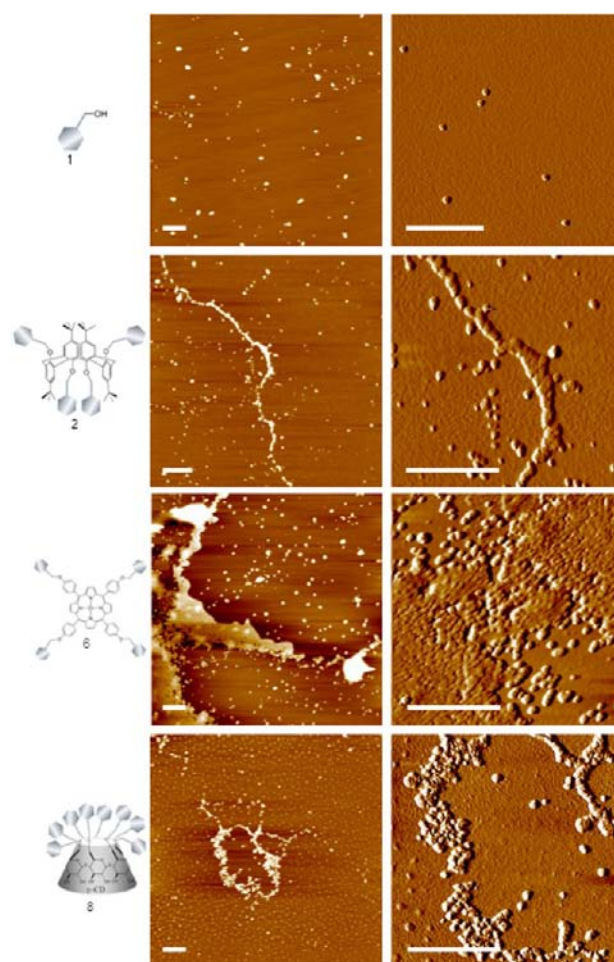


Figure 2. AFM height (left) and deflection (right) images of JB α Man interacting with multivalent ligand **2**, **6**, or **8** ($z = 10$ nm in height images; scale bars = 500 nm).

2, which are paired on each sides of the rigid calixarene scaffold, seems to favor the formation of rectilinear structures compared to more packed aggregates. It is worth mentioning that supramolecular assemblies of similar shapes were previously observed with a bacterial lectin and a tetragalactosylated glycocluster based on the same 1,3-alternate calix[4]arene scaffold.⁴⁰

Significantly different networks were observed with the other multivalent DNJ glycoclusters. Compounds **3**, **4**, **5**, or **8** formed large ring-shaped structures, while the porphyrin-based compound **6**, displaying the highest inhibitory activity, formed much larger aggregates with JB α Man (Figure 2 and Figures S8 and S9, Supporting Information).

Although the supramolecular ring-shape adopted by most of the compounds is difficult to explain by molecular level considerations, the very large 6-JB α Man aggregates formed may be due, at least in part, to the well-known tendency of porphyrin to self-assemble in water through intermolecular π – π stacking interactions.⁴¹ Aggregates of **6** with enhanced valency could play the role of a template facilitating the formation of the larger JB α Man aggregates observed by AFM. Self-assembling multivalent ligands have been previously reported as powerful tools to target bacterial lectins.⁴² This assumption is supported by AFM images of **6** alone showing small supramolecular structures with a thickness of 0.8 to 1 nm (Figure S10, Supporting Information) and by DLS experiments

(Figure S12, Supporting Information, pick around 1 nm). It should be noted that aggregates in Figure 2 have a thickness matching the protein size (5 nm) and are therefore unambiguously ascribed to JB α Man and not to the ligands alone.

AFM analyses were also performed with different 6/JB α Man ratios. The AFM images showed that the size of aggregates increases with growing 6/JB α Man ratio (Figure S11, Supporting Information). This confirms that the ratio influence the size of structures found in AFM images and further supports the thesis that JB α Man aggregates are promoted by the multivalent iminosugars.

Next, we wanted to investigate if aggregative processes are also observed with glycosidases that do not respond to multivalency. The α -glucosidase (α Glu) and β -galactosidase (β Gal) were selected and imaged with or without 3 and 6 (Figure 3). Aggregates were never observed and coincubated

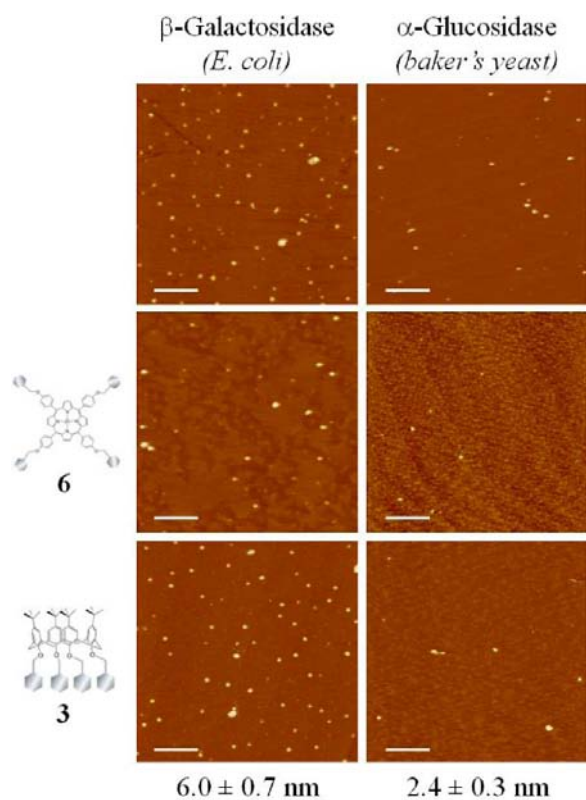


Figure 3. Control AFM images of β -galactosidase and α -glucosidase incubated with tetraivalent ligand 3 or 6 ($z = 10$ nm in height images; scale bar is 500 nm). The mean size of particles is indicated below images. According to these images, these two enzymes cannot form complexes with the multivalent ligands.

solutions showed particles matching the size of α Glu or β Gal alone. These results reinforce the hypothesis that the increased inhibitory activity observed with JB α Man is induced by the formation of large aggregates. This also potentially explains the inversion of the α Glu/JB α Man selectivity observed with the monovalent 1 and the multivalent DNJ inhibitors (Table 1). Despite its preference for the monovalent DNJ motif, α Glu is less inhibited by the multi-DNJ because the enzyme does not respond to multivalency and does not form the large aggregates observed with JB α Man.

The formation of DNJ glycocluster–JB α Man aggregates was further explored in solution by DLS experiments (Table 4 and

Table 4. Hydrodynamic Diameters of JB α Man in the Absence or Presence of Iminosugars^a

JB α Man alone	10.4 \pm 2.2
JB α Man/1	9.9 \pm 2.5
JB α Man/3	663.3 \pm 128.9
JB α Man/6	731.9 \pm 161.9
β -galactosidase (<i>E. coli</i>) /6	7.4 \pm 1.4
α -glucosidase (baker's yeast) /6	3.3 \pm 0.7

^aResults expressed in nm correspond to the mean of three independent samples. The JB α Man/ligand ratio was 1/10. The size of ligands without proteins was less than 1 nm and control proteins (α Glu or β Gal) did not yield aggregates in solution with the ligands.

(Figure S12, Supporting Information). DLS was used to determine the hydrodynamic diameter of JB α Man alone or in complex with monovalent compound 1 and tetraivalent compounds 3 and 6, both showing the highest inhibitory activity with the enzyme. A hydrodynamic diameter of 10 nm was observed for JB α Man alone. Co-incubation with the monovalent reference 1 led to a similar particle-size distribution. In stark contrast, large aggregates of more than 500 nm were formed in the presence of the tetraivalent iminosugars 3 or 6. These results are in good agreement with the AFM studies and further confirm the potency of the tetraivalent iminosugars to form cross-linked complexes with JB α Man.

CONCLUSION

After our first proof of the concept of the multivalent inhibition of JB α Man,⁷ we wished to gain more insight into the binding mode operating with this enzyme. The different affinities and inhibitory profiles observed here with compounds of identical valencies, clearly highlight that the spatial presentation of the DNJ epitopes modulates the strength of the multivalent effects to a large extent. A careful selection of the scaffold is therefore essential to provide an adequate spatial distribution of the DNJ ligands. The octavalent compounds with the highest valencies were not the strongest inhibitors, and this further suggests that designing multivalent iminosugars with fine-tuned spatial distribution of the DNJ is much more relevant than an ever-increasing valency. The porphyrin-based compound 6, displayed a high JB α Man selectivity, with a dramatic 800-fold improvement compared to the monovalent reference, which is unprecedented for a tetraivalent ligand. Insights into the multivalent binding mode operating are also provided here. DLS experiments indicate that large JB α Man aggregates are formed in the presence of the tetraivalent compounds 3 and 6 but not with the monovalent 1. AFM images and DLS measurements also clearly show that the multivalent compounds are able to aggregate dimeric JB α Man, but not α Glu or β Gal which did not respond to multivalency in the inhibitory assay, giving a rational basis for the multivalent effect seen with JB α Man. Interestingly, different patterns are observed depending on the multivalent inhibitors, and this may well explain the difference in inhibition among the compounds. Furthermore, JB α Man is a class II- α -mannosidase of the GH 38 family, a group of glycoside hydrolases of considerable interest, including GM and LM. Inhibitory activities measured with 6 on the pharmaceutical target GM showed an increased affinity

compared to the monovalent reference **1** with a high selectivity over LM. This finding is of particular interest considering the importance of selectively inhibiting GM over LM in cancer chemotherapy. Studies are currently in progress in our laboratories to evaluate the potential of new multivalent iminosugars for selective GM inhibition.

■ ASSOCIATED CONTENT

📄 Supporting Information

Experimental procedures, AFM images, and spectral data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was carried out with financial support from the Centre National de la Recherche Scientifique and the Ministère Délégué à l'Enseignement Supérieur et à la Recherche. Dr F. Albrieux, C. Duchamp and N. Henriques from the Centre Commun de Spectrométrie de Masse (University of Lyon) are gratefully acknowledged for mass spectrometry analyses. We thank Prof. I. B. H. Wilson (Universität für Bodenkultur, Wien, Austria) for generously providing the GMIIb plasmid. The Spanish Ministerio de Economía y Competitividad (Contract SAF2010-15670), the Fondo Europeo de Desarrollo Regional (FEDER), the Fondo Social Europeo (FSE), the Fundación Ramón Areces, and the Junta de Andalucía are gratefully acknowledged for funding.

■ REFERENCES

- (1) For an overview, see: (a) Deniaud, D.; Julienne, K.; Gouin, S. G. *Org. Biomol. Chem.* **2011**, *9*, 966–979. (b) Pieters, R. J. *Med. Res. Rev.* **2007**, *27*, 796–816. (h) Pieters, R. J. *Org. Biomol. Chem.* **2009**, *7*, 2013–2025. (c) Imberty, A.; Chabre, Y. M.; Roy, R. *Chem.—Eur. J.* **2008**, *14*, 7490–7499. (d) Chabre, Y. M.; Roy, R. *Adv. Carbohydr. Chem. Biochem.* **2010**, *63*, 165–393. (e) Dondoni, A.; Marra, A. *Chem. Rev.* **2010**, *110*, 4949–4977. (f) Martínez, Á.; Ortiz-Mellet, C.; García-Fernández, J. M. *Chem. Soc. Rev.* **2013**, *42*, 4746–4773. (g) Bernardi, A.; Jiménez-Barbero, J.; Casnati, A.; De Castro, C.; Darbre, T.; Fieschi, F.; Finne, J.; Funken, H.; Jaeger, K.-E.; Lahmann, M.; Lindhorst, T. K.; Marradi, M.; Messner, P.; Molinaro, A.; Murphy, P. V.; Nativi, C.; Oscarson, S.; Penadés, S.; Peri, F.; Pieters, R. J.; Renaudet, O.; Raymond, J.-L.; Richichi, B.; Rojo, J.; Sansone, F.; Schäffer, C.; Turnbull, W. B.; Velasco-Torrijos, T.; Vidal, S.; Vincent, S.; Wennekes, T.; Zuilhof, H.; Imberty, A. *Chem. Soc. Rev.* **2013**, *42*, 4709–4727.
- (2) (a) Lee, Y. C.; Lee, R. T. *Acc. Chem. Res.* **1995**, *28*, 321–327. For a review, see: (b) Lundquist, J. J.; Toone, E. J. *Chem. Rev.* **2002**, *102*, 555–578.
- (3) For a review, see: Mammen, M.; Choi, S. K.; Whitesides, G. M. *Angew. Chem., Int. Ed.* **1998**, *37*, 2754–2794.
- (4) Kitov, P. I.; Sadowska, J. M.; Mulvey, G.; Armstrong, G. D.; Ling, H.; Pannu, N. S.; Read, R. J.; Bundle, D. R. *Nature* **2000**, *403*, 669–672.
- (5) (a) Lohse, A.; Jensen, K. B.; Lundgren, K.; Bols, M. *Bioorg. Med. Chem.* **1999**, *7*, 1965–1971. (b) John, B. A.; Johnson, C. R. *Tetrahedron Lett.* **1998**, *39*, 749–752. (c) McCort, I.; Sanière, M.; Le Merrer, Y. *Tetrahedron* **2003**, *59*, 2693–2700.
- (6) (a) Benito, J. M.; Gómez-García, M.; Ortiz-Mellet, C.; Baussanne, I.; Defaye, J.; García-Fernández, J. M. *J. Am. Chem. Soc.* **2004**, *126*, 10355–10363. (b) Quesenberry, M. S.; Lee, R. T.; Lee, Y. C.

- Biochemistry* **1997**, *36*, 2724–2732. (c) Dam, K.; Gerken, T. A.; Kavada, B. S.; Nascimento, K. S.; Moura, T. R.; Brewer, F. C. *J. Biol. Chem.* **2007**, *282*, 28256–28263. (d) Gouin, S. G.; Vanquelef, E.; García-Fernández, J. M.; Ortiz-Mellet, C.; Dupradeau, F.-Y.; Kovensky, J. *J. Org. Chem.* **2007**, *72*, 9032–9045. (e) Gouin, S. G.; García Fernández, J. M.; Vanquelef, E.; Dupradeau, F.-Y.; Ortégaz-Munoz, M.; Leffler, H.; Kovensky, J. *ChemBioChem* **2010**, *11*, 1430–1442.
- (7) Diot, J.; García-Moreno, I.; Gouin, S. G.; Ortiz-Mellet, C.; Haupt, K.; Kovensky, J. *Org. Biomol. Chem.* **2009**, *7*, 357–363.
- (8) Compain, P.; Decroocq, C.; Iehl, J.; Holler, M.; Hazelard, D.; Mena Barragán, T.; Ortiz-Mellet, C.; Nierengarten, J.-F. *Angew. Chem., Int. Ed.* **2010**, *49*, 5753–5756.
- (9) Decroocq, C.; Rodríguez-Lucena, D.; Ikeda, K.; Asano, N.; Compain, P. *ChemBioChem* **2012**, *13*, 661–664.
- (10) Durka, M.; Buffet, K.; Iehl, J.; Holler, M.; Nierengarten, J.-F.; Vincent, S. P. *Chem.—Eur. J.* **2012**, *18*, 641–651.
- (11) (a) Dennis, J. W.; Koch, K.; Yousefi, S.; Van der Elst, I. *Cancer Res.* **1990**, *50*, 1867–1872. (b) Goss, P. E.; Reid, C. L.; Bailey, D.; Dennis, J. W. *Clin. Cancer Res.* **1997**, *3*, 1077–1086.
- (12) Goss, P. E.; Baker, M. A.; Carver, J. P.; Dennis, J. W. *Clin. Cancer Res.* **1995**, *1*, 935–944.
- (13) Ju, Y.; Kumar, D.; Varma, R. S. *J. Org. Chem.* **2006**, *71*, 6697.
- (14) Matos, C. R. R.; Lopes, R. S. C.; Cristobal Lopes, C. *Synthesis* **1999**, *4*, 571.
- (15) Dondoni, A.; Marra, A. *J. Org. Chem.* **2006**, *71*, 7546–7557.
- (16) Diot, J. D.; Garcia Moreno, I.; Twigg, G.; Ortiz Mellet, C.; Haupt, K.; Butters, T. D.; Kovensky, J.; Gouin, S. G. *J. Org. Chem.* **2011**, *76*, 7757–7768.
- (17) Goddard-Borger, E. D.; Stick, R. V. *Org. Lett.* **2007**, *9*, 3797–3800.
- (18) Alper, P. B.; Hung, S.-C.; Wong, C.-H. *Tetrahedron Lett.* **1996**, *34*, 6029–6032.
- (19) Chetcuti, M. J.; Devoille, A. M. J.; Othman, A. B.; Souane, R.; Thuery, P.; Vicens, J. *Dalton Trans.* **2009**, 2999–3008.
- (20) Ryu, E. H.; Zhao, Y. *Org. Lett.* **2005**, *7*, 1035–1037.
- (21) Perez-Balderas, F.; Morales-Sanfrutos, J.; Hernandez-Mateo, F.; Isac-García, J.; Santoyo-Gonzalez, F. *Eur. J. Org. Chem.* **2009**, 2441–2453.
- (22) (a) Cecioni, S.; Faure, S.; Darbost, U.; Bonnamour, I.; Parrot-Lopez, H.; Roy, O.; Taillefumier, C.; Wimmerova, M.; Praly, J.-P.; Imberty, A.; Vidal, S. *Chem.—Eur. J.* **2011**, *17*, 2146–2159. (b) Liu, Y.; Ke, C.-F.; Zhang, H.-Y.; Cui, J.; Ding, F. *J. Am. Chem. Soc.* **2008**, *130*, 600–605. (c) Chen, Y.; Vedala, H.; Kotchey, G. P.; Audfray, A.; Cecioni, S.; Imberty, A.; Vidal, S.; Star, A. *ACS Nano* **2012**, *6*, 760–770. (d) Vedala, H.; Chen, Y.; Cecioni, S.; Imberty, A.; Vidal, S.; Star, A. *Nano Lett.* **2011**, *1*, 170–175.
- (23) Rodios, N. A. *J. Heterocycl. Chem.* **1984**, *21*, 1169–1173.
- (24) For an example, see: André, S.; Sansone, F.; Kaltner, H.; Casnati, A.; Kopitz, J.; Gabius, H.-J.; Ungaro, R. *ChemBioChem* **2008**, *9*, 1649–1661.
- (25) (a) Vrasidas, I.; André, S.; Valentini, P.; Böck, C.; Lensch, M.; Kaltner, H.; Liskamp, R. M. J.; Gabius, H.-J.; Pieters, R. J. *Org. Biomol. Chem.* **2003**, *1*, 803–810. (b) André, S.; Liu, B.; Gabius, H.-J.; Roy, R. *Org. Biomol. Chem.* **2003**, *1*, 3909–3916.
- (26) Cecioni, S.; Lalor, R.; Blanchard, B.; Praly, J.-P.; Imberty, A.; Matthews, S. E.; Vidal, S. *Chem.—Eur. J.* **2009**, *15*, 13232–13240.
- (27) For other imino-sugar ligands based on calixarene cores, see: (a) Cardona, F.; Isoldi, G.; Sansone, F.; Casnati, A.; Goti, A. *J. Org. Chem.* **2012**, *77*, 6980–6988. (b) Marradi, M.; Cicchi, S.; Sansone, F.; Casnati, A.; Goti, A. *Beilstein J. Org. Chem.* **2012**, *8*, 951–957.
- (28) Wennekes, T.; van den Berg, R. J. B. H. N.; Bongers, K. M.; Donker-Koopman, W. E.; Ghisaidoobe, A.; van der Marel, G. A.; Strijland, A.; Aerts, J. M. F. G.; Overkleeft, H. S. *Tetrahedron: Asymmetry* **2009**, *20*, 836–846.
- (29) Snaith, S. M. *Biochem. J.* **1975**, *147*, 83–90.
- (30) We used the Quantofic copper test stick for a recent example in the context of CuAAC; see: Baron, A.; Blériot, Y.; Sollogoub, M.; Vauzeilles, B. *Org. Biomol. Chem.* **2008**, *6*, 1898–1901.

(31) Decroocq, C.; Rodríguez-Lucena, D.; Russo, V.; Mena Barragán, T.; Ortiz-Mellet, C.; Compain, P. *Chem.—Eur. J.* **2011**, *17*, 13825–13831.

(32) (a) Kanai, M.; Mortell, K. H.; Kiessling, L. L. *J. Am. Chem. Soc.* **1997**, *119*, 9931–9932. (b) Ashton, P. R.; Hounsell, E. F.; Jayaraman, N.; Nilsen, T. M.; Spencer, N.; Stoddart, J. F.; Young, M. *J. Org. Chem.* **1998**, *63*, 3429–3437. (c) Pagé, D.; Roy, R. *Bioconjugate Chem.* **1997**, *8*, 714–723.

(33) Nemčovičová, I.; Šesták, S.; Rendić, D.; Plšková, M.; Mucha, J.; Wilson, I. B. H. *Glycoconjugate J.* **2013**, DOI: 10.1007/s10719-013-9495-5.

(34) Howards, S.; He, S.; Withers, S. G. *Biochem. J.* **1998**, *273*, 2067–2072.

(35) Kumar, A.; Gaikwad, S. M. *Int. J. Biol. Macromol.* **2011**, *49*, 1066–1071.

(36) Almant, M.; Mastouri, A.; Gallego-Yerga, L.; García-Fernández, J. M.; Ortiz-Mellet, C.; Kovensky, J.; Morandat, S.; El Kirat, K.; Gouin, S. G. *Chem.—Eur. J.* **2013**, *19*, 729–736.

(37) Dam, T. K.; Gerken, T. A.; Brewer, C. F. *Biochemistry* **2009**, *48*, 3822–3827.

(38) Dam, T. K.; Gerken, T. A.; Kavada, B. S.; Nascimento, K. S.; Moura, T. R.; Brewer, F. C. *J. Biol. Chem.* **2007**, *282*, 28256–28263.

(39) (a) Sacchettini, J. C.; Baum, L. G.; Brewer, F. C. *Biochemistry* **2001**, *40*, 3009–3015. (b) Brewer, F. C.; Micelli, M. C.; Baum, L. G. *Curr. Opin. Struct. Biol.* **2002**, *12*, 616–623.

(40) Sicard, D.; Cecioni, S.; Iazykov, M.; Chevolot, Y.; Matthews, S. E.; Praly, J.-P.; Souteyrand, E.; Imberty, A.; Vidal, S.; Phaner-Goutorbe, M. *Chem. Commun.* **2011**, *47*, 9483–9485.

(41) Kiba, T.; Suzuki, H.; Hosokawa, K.; Kobayashi, H.; Baba, S.; Kakuchi, T.; Sato, S.-i. *J. Phys. Chem. B* **2009**, *113*, 11560–11563.

(42) Muller, M. K.; Brunsveld, M. *Angew. Chem. Int. Ed* **2009**, *48*, 2921–2924.